NEW POLYBROMINATED DIPHENYL ETHER FROM THE MARINE SPONGE Dysidea herbacea

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Tropical marine sponges of the family Dysideidae contain a series of antimicrobial polybrominated diphenyl ethers with varying degrees of hydroxylation, methoxylation, and bromination [1]. Recently this class of metabolites has attracted attention because of their ability to inhibit 15-lipoxygenase in mammals [2] and the assembly of microtubular proteins [3] and to regulate interleukin-8 production [4].

We investigated the marine sponge *Dysidea herbacea* collected in the Large Barrier Reef during the ninth cruise of R/V Academic Oparin. Sponge (3 g) was lyophilized and extracted with CHCl₃. The extract was repeatedly chromatographed over silica gel to produce **1** (0.9 mg), **2** (10 mg), **3** (4.5 mg), **4** (0.6 mg), **5** (0.5 mg), and **6** (0.5 mg). Compounds **2-6** were identified by comparing their spectral properties with those in the literature [5, 6]. Diphenyl ether **1**, mp 181-183°C (CHCl₃) was a new representative of this class.

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1: $R = R_1 = X = Y = H$

2: $R = R_1 = H$, X = Y = Br

3: $R = R_1 = Y = H$, X = Br

4: $R = R_1 = Me$, X = Y = Br

5: R = Me, $R_1 = Y = H$, X = Br

6: R = X = Y = H, $R_1 = Me$

The mass spectrum of 1 showed a molecular ion as a cluster of peaks at m/z 522, 520, 518, 516, and 514 and a molecular weight of 517.7898, which corresponded with the empirical formula $C_{12}H_6Br_4O_3$. The PMR spectrum had two pairs of protons with a *meta*-coupling constant. Each pair of protons was determined by selective decoupling. The ¹³C NMR spectrum showed signals for 12 C atoms. C atoms bound to protons were determined by HSQC. The presence of two pairs of *meta*-coupled protons suggested that the molecule had two asymmetric 1,2,3,5-tetrasubstituted aromatic rings.

Comparison of the ¹³C NMR spectrum of **1** with those of **6**, **2**, and **3** showed that the chemical shifts of C-1 to C-6 of **1** corresponded with those of C atoms in ring A of **6** [7]; those of C-1' to C-6', of C atoms of ring B in **2** and **3** [5] (Table 1). C atoms in **1** were assigned using HMBC. Methylation of **1** by MeI gave a dimethyl ether, the spectral properties of which agreed with the literature data for 2-(3',5'-dibromo-2'-methoxyphenoxy)-3,5-dibromoanisole [5]. Thus, **1** had the structure 2-(3',5'-dibromo-2'-hydroxyphenoxy)-3,5-dibromophenol.

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TABLE 1. ¹³C NMR (125 MHz), PMR (500 MHz), and HMBC Spectra of **1** (acetone-d₆)

C atom	$\delta_{ m C}$	$\delta_{H}\left(J/Hz\right)$	НМВС
1	152.6		
2	139.3		
3	118.6		
4	126.8	7.38 d (J = 2.1)	2, 3, 5, 6
5	119.6		
6	120.7	7.24 d (J = 2.1)	1, 2, 4, 5
1'	146.7		
2'	144.5		
3'	111.2		
4'	129.1	7.37 d (J = 2.1)	2', 3', 5', 6'
5′	110.7		
6'	116.3	6.64 d (J = 2.1)	1', 2', 4', 5'

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REFERENCES

- 1. D. J. Faulkner, *Nat. Prod. Rep.*, **19**, 1 (2002) and preceding articles.
- 2. E. N. Segraves, R. R. Shah, N. L. Segraves, T. A. Johnson, S. Whitman, J. K. Sui, V. Kenyon, R. H. Cichewicz, P. Crews, and T. R. Holman, *J. Med. Chem.*, **47**, 4060 (2004).
- 3. H. Liu, M. Namikoshi, S. Meguro, H. Nagai, H. Kobayashi, and X. Yao, J. Nat. Prod., 67, 472 (2004).
- 4. T. Oda, H. Liu, and M. Namikoshi, *Mar. Drugs*, **3**, 119 (2005).
- 5. R. S. Norton, K. D. Croft, and R. J. Wells, *Tetrahedron*, **37**, 2341 (1981).
- 6. X. Fu, F. J. Schmitz, M. Govindan, S. A. Abbas, K. M. Hanson, P. A. Norton, P. Crews, M. Laney, and R. C. Schatzman, *J. Nat. Prod.*, **58**, 1384 (1995).
- 7. N. K. Utkina, M. V. Kazantseva, and V. A. Denisenko, Khim. Prir. Soedin., 603 (1987).